Copper Solvent Extraction: Status, Operating Practices and Challenges in the African Copper Belt

Kathryn C. Sole\textsuperscript{1} and Owen Tinkler\textsuperscript{2}

\textsuperscript{1}Consulting Hydrometallurgist, Johannesburg, South Africa
\textsuperscript{2}Cytec Industries, Phoenix, AZ, United States of America

*Corresponding author: kathysole09@gmail.com

Although the first large-scale application of copper recovery by solvent extraction (SX) took place in Zambia in the early 1970s, it is only in the last decade that this technology has become widely employed in this part of the world and is now a mainstay unit operation in most copper hydrometallurgical flowsheets. The mineralogy of the ores in the African Copper Belt, and hence the characteristics of the African leach liquors, differ significantly from those in Chile and the south-western USA, where copper SX has had a long and successful history. These differences provide operators, metallurgists, reagent vendors and engineers with many challenges: new approaches are needed to adapt this technology for successful implementation in this region. This paper examines typical operating practice in the African Copper Belt, discusses differences compared with other parts of the world and looks at some of the challenges and opportunities presented by these flowsheets.

INTRODUCTION

Following the success of the Rancher’s Bluebird and Bagdad solvent extraction and electrowinning (SX-EW) operations in Arizona in the late 1960s, the Tailings Leach Plant at Chingola, Zambia became the first large-scale copper SX plant in the world, commissioned in 1974. Despite the equipment design now being outdated, this plant still continues to operate successfully, indicating the versatility and adaptability of this technology. Today, there are some 75 copper SX operations worldwide with cathode production above 10 kt/a. The top ten producers currently account for some 40\% of the global 4.3 Mt/a SX-EW copper production. South America (predominantly Chile and Peru) is the largest copper cathode-producing region, with annual production of some 2 Mt. The Central African Copper Belt (Zambia and the Democratic Republic of Congo (DRC)) is second, with cathode production of 1.2 Mt/a and North America (USA and Mexico) third, producing close to 0.8 Mt/a Cu (Cytec data, 2014). The remaining 0.35 Mt/a comes from all other regions combined (referred to as “Rest-of-World”).

Current and future operations in Zambia and the DRC are summarised in Figure 1. The industry in this region is characterised by a large number of small (3 to 20 kt/a) plants, mainly Chinese-owned, and a few very large (>200 kt/a) operations, owned by major multinational corporations.

This paper examines typical operating practice in the African Copper Belt, discusses the differences compared with other parts of the world and looks at some of the challenges presented by the conditions, as well as innovations in flowsheets, reagents and operating conditions that have been introduced in these circuits.
Figure 1. Operating copper SX–EW plants and near-term projects in the Copper Belt.

COPPER SX: A TRULY ADAPTABLE PROCESS

Distinct characteristics emerge when one examines copper SX on a regional basis. Table I compares characteristics of the pregnant leach solutions (PLS) in Central Africa with those of North America, South America, and other parts of the world where copper SX is practiced. The nature of the PLS varies considerably (0.23 to 43 g/L Cu), depending largely on the type of process and geographic location (Sole et al., 2013). The versatility of this technology is indicated by its ability to cope with extreme variations in copper grade as well as selectively recover copper from widely varying PLS compositions. Extreme examples include Mexicana de Cananea (Mexico), that processes a feed containing 2 g/L Cu and 45 g/L Fe, and Cobre Las Cruces (Spain) which recovers >40 g/L Cu from a PLS background of >50 g/L Fe.

Table I. Variation of average PLS composition with location.

<table>
<thead>
<tr>
<th>Location</th>
<th>Cu (g/L)</th>
<th>pH</th>
<th>Fe (g/L)</th>
<th>Mn (g/L)</th>
<th>Co (g/L)</th>
<th>Cl (mg/L)</th>
<th>TSS (mg/L)</th>
</tr>
</thead>
<tbody>
<tr>
<td>North America</td>
<td>0.3 – 3</td>
<td>1.5 – 2.5</td>
<td>3 – 45</td>
<td>1 – 3</td>
<td>–</td>
<td>70</td>
<td>30</td>
</tr>
<tr>
<td>South America</td>
<td>1.0 – 6.5</td>
<td>1.1 – 2.5</td>
<td>0.3 – 31</td>
<td>0.3 – 12</td>
<td>0.06</td>
<td>870</td>
<td>44</td>
</tr>
<tr>
<td>Africa</td>
<td>3.0 – 43</td>
<td>1.5 – 2</td>
<td>3.4</td>
<td>0.9</td>
<td>6.7</td>
<td>–</td>
<td>137</td>
</tr>
<tr>
<td>Rest-of-World</td>
<td>1.0 – 40</td>
<td>1.0 – 2.5</td>
<td>–</td>
<td>–</td>
<td>–</td>
<td>–</td>
<td>–</td>
</tr>
</tbody>
</table>

Copper grades are typically lowest in the Southwestern USA, where the industry is mature and many operations have been in production since the mid-1980s. The ores are mainly lower-grade mixed oxide and secondary sulphide, containing 0.2 to 0.3% acid-soluble copper. The operations are typically heap and dump leaching, with permanent pads (rather than on-off pads) to minimise operational costs. An acid cure is common to increase leach recovery. The SX circuits are often configured for series-parallel or all-parallel operation, as this maximises copper production by treating high volumes of low-grade PLS. At most sites, the high PLS flowrates give extraction organic-to-aqueous (O:A) flowrate ratios well below 1:1, so raffinate stages operate in aqueous phase continuity—a feature that is unusual in other parts of the world. These operations are characterised by efforts to minimise operating costs and extend the profitable life of mine for as long as possible. There is an emphasis on minimising organic losses and maximising organic recovery (e.g., by increased retention time in the raffinate pond and use of equipment such as Pace Setters, Jameson cells and pond skimmers).
Chilean operations process mainly oxides with an acid-soluble copper grade of 0.4 to 0.8%. Many of these are located in the Atacama Desert, where the predominant mineral, atacamite (Cu₂Cl(OH)₃), gives levels of chloride in the PLS as high as 40 g/L. The presence of >30 mg/L Cl is detrimental to copper EW (Lakshmanan et al., 1977), so SX circuits usually have a wash stage to limit chloride transfer to the electrolyte. Several sites also have high levels of nitrates in the PLS. Nitrate is a strong oxidising agent, which presents severe challenges with accelerated reagent degradation. Heap leaching of primary and secondary sulphides is becoming more common; primary sulphides frequently give elevated levels of Fe in the PLS so reagent selectivity is particularly important. Processing is mainly by heap, dump and run-of-mine (ROM) leaching.

The majority of Australian leach-SX-EW operations have come to the end of their operating lives, so the Rest-of-World region is now characterised by expansion in countries such as Laos, Myanmar, Kazakhstan, and China. Olympic Dam, one of last remaining Australian operations, has complex metallurgy (copper is produced as a byproduct of uranium) and a life of mine of at least 200 years (Russell, 2014). The Rest-of-World locations are often very remote, with difficult logistics. Some of the flowsheets are complex, comprising pressure leaching as well as agitated tank leaching, and there is a wide variation in the types of leach solutions that are processed by SX.

The African operations have the advantage of high ore grades (3 to 5% acid-soluble Cu), and, consequently, significantly higher PLS copper grades than found in other regions (43 g/L Cu design at Kamoto Copper Company, for example). Although the predominant oxide minerals, such as malachite, chrysocolla, and heterogenite, are easily leachable, acid consumption varies dramatically from site to site and even within orebodies—in some cases, rendering even high-grade ore bodies uneconomical. After crushing and grinding, leaching is usually carried out at slightly elevated temperature (35 to 45°C) in agitated tanks, which results in dissolution of almost all the acid-soluble copper in a matter of hours, rather than months or even years in the case of heap leaching.

An additional feature of DRC operations is that the ore bodies often contain significant quantities of cobalt, usually produced as an intermediate cobalt hydroxide product which is further refined in Europe or China. A few operations produce cobalt cathode after solution purification. Logistics in this region remain very difficult, with most reagents having to be transported by road from South Africa, Namibia, or Tanzania. Many sites have therefore installed sulphur-burning plants to produce their own sulphuric acid on site from elemental sulphur.

It is evident from the above discussion that copper SX is practiced very differently in different parts of the world and that each region has its own unique challenges which require innovation and adaptability to ensure the consistent and profitable production of high-quality copper cathode.

THE COPPER SX-EW LANDSCAPE IN CENTRAL AFRICA

Zambia

Tailings Leach Plant: A World First

The first large-scale copper SX plant in the world was commissioned at the Nchanga Tailings Leach Plant (TLP) in Zambia in 1974, with a daily production of 200 t Cu cathode (Holmes et al., 1976). The mixer-settlers, originally configured as four trains of three extract and two strip stages, are long and narrow (aspect ratio of 2.9:1), while the mixers have a design residence time of 3 min. Given the constraints of the extractants at the time that TLP was commissioned, multiple stages of extract and strip were necessary to achieve the desired copper recovery. With advances in mixer-settler design and extractant formulations, the largest modern-day mixer-settlers could process about 85% of the TLP’s total PLS flow in a single 2E+1S (two extract and one strip stage) circuit!
Largely unknown in the history of African metallurgy is that the first aldoxime reagents (the ACORGA P-5000 Series) were developed by ACORGA Ltd, which was a joint venture between Anglo American Corporation of South Africa and the UK’s Imperial Chemicals Industries (ICI) (a forerunner of what is today the multinational company, Cytec Industries) (Tumilty et al., 1977). The first commercial aldoxime extractants were produced for use in copper SX and exhibited improvements in extracting power, kinetics and selectivity over the existing reagents. Although extractant chemistry, stability and performance have improved significantly since that time, aldoximes (with a stripping modifier) still form the basis of some 66% of copper cathode production today.

**Development of the Industry & Current Operations**

Located 10 km from Ndola, the Bwana Mkubwa SX–EW plant (BMML) was established by First Quantum Minerals (FQM) in 1998 to process oxide tailings dams in the area. In 2003, the processing plant was redesigned and expanded to process oxide ore from the Lonshi deposit, which was located in the DRC, some 35 km away. At its peak in 2005, BMML produced just under 50 kt cathode (FQM, 2006). First Quantum halted operations at BMML between 2008 and 2010 and reopened for about a year in 2010. Although this plant no longer operates, it provided a pioneering example to the industry of how rapidly, inexpensively and simply copper SX flowsheets could be installed and run profitably.

The Kansanshi Mine, 80% owned by FQM subsidiary Kansanshi Mining PLC, reached commercial production in April 2005. It is the largest copper mine in Africa and the 8th largest in the world, with capacity of 400 kt/a Cu (210 kt by SX–EW) and, unusually for this region, 130 000 oz/a Au (FQM, 2015). Ore treatment is flexible to allow for variation in ore type, with processing possible through an oxide circuit, a sulphide circuit and a transitional ore “mixed float” circuit. Sulphide ore is concentrated by crushing, milling and flotation (Chongo & Ngulube, 2015). Oxide ore is treated via crushing, milling, flotation, leaching and SX–EW to produce sulphidic and gold-bearing flotation concentrates, as well as electrowon cathode copper (FQM, 2015). The hydrometallurgical circuit employs both pressure and atmospheric leaching steps, and has five SX circuits and three tankhouses. The site also has its own dedicated acid plants. The autoclave leach results in relatively high levels of iron(III) reporting to the PLS. It is well known that ferric iron deteriorates current efficiency in EW and this plant has battled with transfer of iron to the electrolyte (Mwale & Megaw, 2011).

Mopani Copper Mines (MCM) produced 212 kt of copper (including refined copper from third parties) in 2013, representing a 13% year-on-year increase. The Nkana site has two SX circuits currently operating, one treating the PLS from an oxide leach and the other being used to remove copper in the cobalt circuit. The Mufulira site has three SX circuits that treat a blend of heap leach and vat leaching solutions. Vat leaching is unusual in modern flowsheets—Mantos Blancos in Chile is the only other plant currently employing vat leaching (Schlesinger et al., 2011). Several mining capital projects (Synclinorium Shaft at Nkana, Mufulira Deeps, Mindola Deeps) are coming on line before the end of 2016, which will extend mine life by 20 to 25 years (ZCCM, 2014).

Chambishi Metals, located near Chambishi, is one of the oldest plants in the region. It was commissioned in 1978 and privatised in 1998. In one of the more interesting flowsheets, the plant treats Cu/Co sulphide concentrates imported from DRC via a roast–leach–SX–EW process. The newly built SX circuit ensures the delivery of a low copper tenor feed solution to the cobalt refining circuit as well as the production of high-purity copper cathode. The cobalt purification process starts with lime precipitation followed by SX and ion-exchange processes for zinc and nickel removal, respectively, before cobalt is electrowon, degassed, and crushed to produce LME-grade cobalt metal.

China Non-Ferrous Metals Company (CNMC) started the Sinometals leach plant in 2008. Located near Kitwe, this plant processes a variety of copper-containing raw materials. Through its subsidiary Luanshya Copper Mines (LCM), CNMC also operates the Muliashi SX–EW operation which was commissioned in 2012. Muliashi produces cathode (approximately 40 kt/a) from both agitated-leach and heap-leach circuits.
Challenges & Future Prospects

Until 2012, Zambia was the largest copper producer in Africa, but it has since been overtaken by the DRC. Zambian mining policy has travelled the path from assets being held by privately owned companies, through nationalisation during the 1960s, and back to private ownership again in 2000 (Limpitlaw, 2011). The lack of success of the nationalisation policies has been strongly articulated by the Minister of Mines (Kapembwa, 2014): it is now well accepted that the economy, productivity and health and safety performance fare far better under corporate management. Today, the Zambian Government retains a minority interest in most of the large projects and mines through its holding company, Zambia Consolidated Copper Mines Investment Holdings plc (ZCCM-IH). Under a law passed in January 2015, Zambia’s mining royalty rate has increased more than three-fold, which is putting enormous pressure on the operating copper mines (Hill, 2014a). It is likely that the new law will be revised during 2015 to prevent multiple mine closures and mass lay-offs.

As in many Southern African countries, electricity supply is an increasing source of concern. The national producer, ZESCO, has 2200 MW capacity and produces some 90% of demand, but is increasingly under pressure as additional mining projects come online. In particular, commissioning of FQM’s Sentinel mine and smelter in 2015, will significantly constrain supply. The national regulator granted an electricity tariff increase of 28% in 2014, but this is currently under judicial review after appeals from the mining industry (Hill, 2014b).

Democratic Republic of Congo

Development of the Industry and Current Operations

The DRC is widely considered to be one of the richest countries in the world with respect to natural resources. It contains commercially viable quantities of some 50 commodities, including one-third of the world’s cobalt and 10% of the world’s copper.

Although hydrometallurgy had been employed in the DRC by Union Minière du Haut Katanga and its successor company, Gecamines, since the 1950s, it is only very recently that SX–EW has become commonplace. Some of the first companies to move back into the country in the early 2000s operated furnaces which produced 97 to 99% blister copper from high-grade copper oxide ores while others built leach–direct EW operations. In both cases, the intention was to minimise capital costs and recoup their investments as fast as possible by producing copper on a short lead time. Cathode produced by direct EW was of poor quality, but facilitated cash flow for further expansions and upgrading. The first large-scale modern SX–EW facility was commissioned in 2008. Since then, the industry has expanded rapidly and there are now some 24 operations producing copper via this route.

One of the largest global SX–EW operations is Tenke Fungurume Mining (TFM), located between the two rural towns of these names. TFM has one of the richest orebodies in the world and huge effort and resources were poured into the development of a flowsheet for this project in the 1970s. At that time, the project was far advanced and the emerging technology of SX was a crucial component. With the outbreak of a series of civil wars in (then) Zaire, all mining projects ceased from the mid-1970s. It is only since about 2004 that the political situation in DRC has stabilised sufficiently for adventurous multinational companies to start reinvestment in this region. Tenke, which started operation in 2009 and expanded in 2013, is today considered the flagship operation of the Copper Belt, with 2014 production of 203 kt Cu and 13.2 kt Co as Co(OH)₂. Under the management of the US-based Freeport McMoRan, it also boasts amongst the best safety records in the region and lowest cash operating costs (USD 1.15/lb net of cobalt credits) (Freeport McMoRan, 2014).

The two other major DRC operations are the Glencore-owned Kamoto Copper Company (KCC) and Mutanda Mining, both located near Kolwezi at the far end of the Copper Belt. KCC’s Luulu plant started production in the 1950s under Union Minière du Haut Katanga and was considered a world-class operation at that time. It fell into disuse and disrepair during the war years, but was restarted in 2008 using the original flowsheet. The process comprises agitated leaching of oxide concentrate and roasted sulphide concentrate followed by precipitation separations and direct EW. The first of three
SX circuits was brought on-line in December 2012 and production today is entirely by SX-EW. The plant now has a design capacity of 300 kt/a Cu and produced 200 kt Cu and 2.78 kt Co cathode in 2014 (Katanga Mining, 2014).

Mutanda Mining started as a greenfield project in 2010. It is located on one of the richest deposits in this country and has gone through several expansions since start-up. Today, this operation, which treats whole ore by agitated leaching and heap leaching, boasts four SX circuits and seven EW tankhouses, with a design capacity of 200 kt/a Cu. 2014 production was 197 kt Cu cathode and 14.4 kt Co as Co(OH)₂ (Fleurette Group, 2015).

Located near KCC, Sicomines is the first large Chinese operation in DRC. Commissioning is expected to start in mid-2015, and the first phase of the project will produce 35 kt/a Cu cathode. The second phase, starting mid-2016, will increase this to 150 kt/a. Production of Co(OH)₂ is expected to also start before the end of 2015. There are several other small Chinese operations in DRC, most producing 12 to 20 kt/a Cu cathode. MKM (La Minière de Kalumbwe Myunga) produces 20 kt/a Cu cathode and 600 t/a Co(OH)₂. Comilu (Compagnie Minière de Luisha) started production in 2015 with a design capacity of 14 kt/a cathode using a heap leaching–SX–EW flowsheet and is currently operating at about 50% of design. CIMCO (Congo International Mining Company) currently produces 500 to 1000 t/m cathode, also via heap leaching–SX–EW. Shituru Mining Company (SMCO), located outside Likasi, was commissioned in 2012. It produces 35 kt/a copper cathode by agitated leach–SX–EW.

The Chinese-owned Mineral & Metals Group (MMG) operates Kinsevere, located near Lubumbashi. This started as a dense-medium separation plant, producing concentrate, but today the flowsheet comprises SX and EW, producing above 70 kt/a copper cathode. The available oxide ore is depleting and feasibility studies are currently underway to process sulphide material in the future.

Ruashi Mining, owned by China’s Jinchuan Group (following the 2011 acquisition of Metorex), is located on the outskirts of Lubumbashi. The operation has a long history of copper and cobalt mining, going as far back as 1911. The modern operations have been in production since 2005. These comprise three open pits and copper is produced by leach–SX–EW following flotation concentration of the oxide ore—an unusual unit operation, but one that is employed in a few DRC flowsheets. Copper recovery is maximised by operating high- and low-grade SX circuits. Ruashi produces its own sulphuric acid and SO₂ to reduce the need to purchase acid and sodium metabisulphite (SMBS) that is required to reductively leach cobalt from heterogenite. Cobalt is produced as Co(OH)₂ by precipitation with MgO. Ruashi currently produces 38 kt/a Cu and 4.4 kt/a Co (Metorex, 2015).

Chemaf’s Usoke plant in Lubumbashi produced 15.2 kt Cu and 2.0 t Co in 2014 (Shalina Resources, 2015). The SX–EW equipment was brought from the old Mount Gordon site in Australia in 2007 when that plant closed and has considerable capacity available (31.5 kt/a). Chemaf originally produced cobalt as CoCO₃ but today it is mainly sold as Co(OH)₂, produced by precipitation with MgO. A cobalt EW pilot facility produces 50 t/a cobalt cathode of 99.9% purity. Their nearby Etoile plant started commissioning in January 2015 and is designed to produce 20 kt/a Cu and 3.8 kt/a Co.

Tiger Resources’ Kipoj Project (SEK) is located between Lubumbashi and Likasi and currently produces 25k t/a copper cathode by heap leach–SX–EW. Following several years of operating a heavy-media separation plant, the SX–EW plant was successfully commissioned in May 2014.

The Roan Tailings Reclamation (RTR) project is currently under construction near Kolwezi and will produce 70 kt/a cathode copper during its first phase of operation, currently targeting Q4 2016 for first metal production. The flowsheet is an agitated leach, followed by SX–EW. 14 kt/a Co will also be produced as Co(OH)₂. In contrast to most other DRC operations, this tailings material contains significant qualities of zinc, which will be removed from the cobalt circuit using SX. Because it treats tailings, mining costs are negligible for this project and operating costs are expected to be very low, providing resilience for the operation in times of depressed commodity prices.
Challenges & Future Prospects

Electrowinning of copper is very energy-intensive (~2.1 MWh/t Cu). Decades of infrastructure decay in DRC have left an electricity supply that is unreliable, both from the viewpoints of availability and stability. Power failures up to six times per day are not uncommon. Many operations supplement their power requirements using diesel generators, which add significantly to operating costs. Fortunately, this situation should improve steadily, as mining companies make investments in the regional power grid. The Congo River is estimated by the United Nations to be able to supply the electricity needs of the entire continent by hydroelectric power (International Rivers, 2015a). The proposed Grand Inga scheme, if all seven phases are eventually implemented, will supply 40 000 MW—twice the generating capacity of the world’s largest hydroelectric plant, the Three Gorges in China (Pearce, 2013). In 2013, Inga produced 115 MW DC, which increased to 250 MW in 2014. The Inga-3 project, is expected to start by 2020 and will eventually supply a further 4800 MW (of which, South Africa will take 2500 MW) (International Rivers, 2015b). Reliable and sufficient electricity supply is vital to the successful revitalization of the economy, and particularly the mining industry.

Transport infrastructure also provides huge challenges in this region. Almost all capital equipment and operating consumables must come in by road transport, either from Namibia, South Africa, or Tanzania, a distance of 2200 to 3000 km. Typical overland transit times range from 10 to 20 days. As a comparison of the impact of transport on operating costs, MMG report that the cost of sulphuric acid at Kinsevere, near Lubumbashi, at USD 460/t, is almost double that at their Sepon operation, situated in a remote location in Laos. There are no operating railways in the Copper Belt, but initiatives are underway to bring in rail transport from the deep water port at Walvis Bay, Namibia, (Njini, 2014), and from Dar es Salaam, Tanzania (The Guardian, 2013). Onerous and expensive visa requirements, excessive landing fees (leading to expensive airfares) and difficult travelling conditions are also an impediment to foreign investment and expertise.

The economic and political stability of the country still present unacceptable risks for many investors. The Congolese have the lowest, or near-lowest, nominal gross domestic product (GDP) per capita in the world. In 2014, DRC had the second-lowest Human Development Index of 187 ranked countries (UNDP, 2014) and is also one of the lowest-ranked countries on the Corruption Perception Index (Transparency International, 2014).

On the positive side, the Copper Belt ore bodies are incredibly rich by global standards: tailings sent to waste are typically of much higher grade than the feeds of many older operations in North and South America—which, nevertheless, are able to operate profitably. There is enormous potential for improved productivity, recovery and throughput in these operations. The lowest labour costs of all copper-producing countries (World Salaries, 2015) give ample opportunities for operations to drive themselves down the cost curve and become competitive in global terms. Mutanda is already one of the five lowest-cost producers in the world.

Significant opportunities also exist to capitalise on being a large-scale producer of cobalt. The demand for cobalt is increasing because its main applications are technology-based and in high-temperature alloys. The price of cobalt is likely to increase more than that of copper in the short- to medium-term.

CENTRAL AFRICAN COPPER SX-EW TECHNICAL CHARACTERISTICS AND CHALLENGES

Agitated Leach Flowsheets

Because of the high grade of the Central African ores and their readily acid-leachable minerals, leaching is generally carried out in agitated tanks, in contrast to the percolation (heap, dump and ROM) leaching of most North and South American operations. The up-side of agitated leaching is that the residence time required is of the order of hours (rather than months or years) and high-grade PLS tenors can be achieved. The down-side, however, is that extensive washing of the leach residue is
required to minimise copper (and cobalt) losses to the tailings solids. Multiple counter-current decantation (CCD) units are required, with the associated need to introduce flocculants and other settling-enhancing chemicals into the PLS, which may detrimentally impact the physical behaviour of the SX organic phase. A further disadvantage of solid–liquid separation by CCD is the high levels of total suspended solids (TSS) that report to the PLS and which greatly increase crud formation and the associated losses of extractant and diluent. In percolation leaching systems, the ore bed acts as a filtration medium, so TSS values are much lower (see Table I).

A unique flowsheet configuration, developed as a consequence of agitated leaching and the large washing requirements, is that of the so-called Split Circuit, the patent for which is held by BASF (Kordosky & Nisbett, 2005; Nisbett et al., 2008). The relatively high copper tenor PLS generated directly from the leaching circuit reports to a high-grade SX circuit, which is operated to maximise copper throughput. The acid-containing raffinate is then used for washing in the CCD circuits. This allows some further copper recovery from any incompletely leached solids and utilisation of the “free” acid that has been generated by SX. The low-grade solution from the final CCD overflow reports to a low-grade SX circuit, which is operated to maximise copper recovery and minimise the amount of copper reporting to the final raffinate. In some flowsheets, a portion of this raffinate is neutralised and discarded, so any associated copper represents a loss to tailings.

**Extractant Choices and Consumptions**

The most widely used extractants used are modified and non-modified aldoxime–ketoxime blends, e.g., LIX984N from BASF Corporation and ACORGA OPT5510 & OPT5540 from Cytec Industries. Several plants in Zambia use modified-aldoxime reagents successfully too.

The extractant concentration employed (Table II) obviously bears a direct relationship to the Cu tenor of the PLS (Table I) and extraction efficiency required. Most of the African operations use significantly higher extractant concentrations than elsewhere: for example, Mutanda employs 35 vol.% to process a PLS containing 22.5 g/L Cu, while KCC uses 33 vol.% for a PLS of 26 g/L Cu.

**Table II. Variation of average extractant and diluent parameters with location (Schlesinger et al., 2011).**

<table>
<thead>
<tr>
<th>Location</th>
<th>Conc. (vol.%)</th>
<th>Net transfer (g/L Cu/vol.%)</th>
<th>Usage (kg/t Cu)</th>
<th>Aromatics (vol.%)</th>
<th>Flash point (°C)</th>
<th>Usage (kg/t Cu)</th>
</tr>
</thead>
<tbody>
<tr>
<td>N. America</td>
<td>8.7</td>
<td>0.19</td>
<td>3.9</td>
<td>12</td>
<td>79</td>
<td>31.4</td>
</tr>
<tr>
<td>S. America</td>
<td>15.4</td>
<td>0.28</td>
<td>2.7</td>
<td>10</td>
<td>80</td>
<td>22.9</td>
</tr>
<tr>
<td>Africa</td>
<td>30.3</td>
<td>0.23</td>
<td>4.1</td>
<td>18</td>
<td>87</td>
<td>12.4</td>
</tr>
</tbody>
</table>

The high TSS levels in agitated leaching PLS are responsible for crud formation in SX circuits. A consequence is that extractant consumption is much higher than would be expected for the processing of high-grade copper liquors. To illustrate this, Table III compares reagent consumption of three hypothetical, well-operated, mid-sized operations in Chile, the USA and the DRC. In this comparison, the assumption is made that organic losses are the same (50 mg/L) and annual production is 46 kt.

Clearly the reagent usage in the African case should be significantly lower than in Chile or the USA. Since actual usages are more in the 3 to 6 kg/t range, the difference is attributable both to higher entrained losses of organic because of crud build-up and because of the significant organic losses associated with crud processing.

Interestingly, although a number of the operations have mixer temperatures in the 35–45 °C range, hydrolytic degradation of the extractant is seldom an issue. This can be attributed to the relatively high replacement or ‘make-up’ rate, as discussed above.
Table III. Comparison of expected and actual extractant consumptions for well-operated plants.

<table>
<thead>
<tr>
<th>Location</th>
<th>Circuit config.</th>
<th>PLS Cu (g/L)</th>
<th>Flow (m$^3$/h)</th>
<th>Extract O/A</th>
<th>ΔCu (PLS-Raff, g/L)</th>
<th>Extractant (vol.%)</th>
<th>Extractant usage (kg/t cathode)</th>
</tr>
</thead>
<tbody>
<tr>
<td>USA</td>
<td>2E+1S</td>
<td>1.55</td>
<td>3500</td>
<td>0.6</td>
<td>1.5</td>
<td>12</td>
<td>3.9</td>
</tr>
<tr>
<td>Chile</td>
<td>2E+2S</td>
<td>5.5</td>
<td>1050</td>
<td>1.0</td>
<td>5</td>
<td>15</td>
<td>1.5</td>
</tr>
<tr>
<td>DRC</td>
<td>2E+2S</td>
<td>12</td>
<td>477</td>
<td>1.5</td>
<td>11</td>
<td>25</td>
<td>1.1</td>
</tr>
</tbody>
</table>

Diluent Choices and Consumptions

Safety considerations, particularly flash point, dominate diluent selection. Despite the growing global tendency to minimise carbon footprint and use aliphatic diluents, in Africa, diluents containing higher aromatic content (8 to 25 vol.%) have been traditionally preferred for the enhanced solvating capacity that this offers for the higher extractant concentrations (Table II). A recent study, however, indicates that aliphatic diluents may also be successfully used for these applications (Brown, 2015).

Of increasing concern to the regional industry is diluent availability. Until recently, all diluent has been supplied from the South African Petroleum Refinery (SAPRef). However, with advancing age and poor maintenance, this refinery is suffering increasingly frequent and lengthy shutdowns (137 days of lost production in 2014), which is aggravating diluent supply. The other South African and Zambian refineries do not produce suitable products for this application and there is an increasing need to import alternatives. Although these offer suitable safety and chemical properties, imports obviously command a significant price premium and add to ever-increasing operating costs and lead times on orders. This issue will not be resolved in the short-term and innovative solutions are required to ensure that the increasing demands of Copper Belt users can be consistently and reliably met.

Suspended Solids and Crud Management

The rate of build-up of crud, both at the settler interface and on the bottom of the settlers, is by far the most significant operational issue facing the African agitated leach–SX operations. To put it in perspective, heap-leach operations typically deliver TSS in the PLS of <30 mg/L, while—even after clarification—most agitated-leach PLS will contain upwards of 50 mg/L and frequently over 100 mg/L TSS. If not actively managed, the crud will literally fill up the settlers over a six- to twelve-month period. In fact, even with regular removal, most agitated leach-SX operations regularly take individual settlers off-line annually to remove bottom crud. Minimising the entry of solids into SX is obviously the best approach, although this is difficult to accomplish in practice. Several operations have installed pinned-bed clarifiers on the PLS streams; these have seldom been effective although there are examples where TSS has been reduced to <20 mg/L.

Crud exacerbates losses of the organic phase and crud-treatment systems are required to reclaim as much of these expensive reagents as possible. It is usually necessary to clay treat any recovered organic phase before returning it to the circuit to avoid introducing organic degradation products or other detrimental compounds into the SX circuit. Although expensive, three-phase centrifuges and plate-and-frame filters have become the standard equipment both to separate the organic phase from the solids and aqueous phase and to separate organic from clay. The solid component of crud ultimately reports to tailings disposal.

Impact of Up- and Downstream Additives

Flocculants, coagulants, flotation chemicals and similar reagents are surfactants. If the addition of these chemicals to upstream unit operations is not properly controlled, they can have severe detrimental impacts on the SX circuit. Because the SX process is interfacial – involving the transfer of
Cu$^{2+}$ ions across an aqueous-organic interface—its chemical and physical performance are strongly influenced by the presence of any foreign species that interfere with the interfacial characteristics. It is critical to evaluate the potential impact that the upstream introduction of a new reagent may have on the downstream SX performance.

The transfer of organics, such as mist suppressants and smoothing agents, from EW to SX via the return spent electrolyte, can also have unintended consequences if not properly managed.

**Impact of Dissolved and Hydrated Silica Species**

African PLS often contain very high levels of dissolved or hydrated silica species. As is well-known, silica can form polymeric structures under appropriate conditions of temperature and aging. These can significantly alter the viscosity of the solutions in the SX circuit, which impedes the transfer of copper across the organic-aqueous interface. Silica is usually also a major component of crud and can contribute to significant crud formation.

Coagulants can be used to remove colloidal silica and fine particles (less than 0.5 µm diameter). Because coagulants are surface-active agents, it is important to ensure that these do not interfere with the interfacial behaviour in the SX circuit. Neutral (uncharged) coagulants, which cannot be overdosed, are generally preferred. Liquid, rather than the commonly used solid powder-based coagulants, are safer to use (no inhalation hazard) and can be used directly without prior dilution. Chambishi has, for many years, employed silica coagulants ahead of both their copper and cobalt SX circuits to remove silica from the PLS and avoid severe clogging of their filters.

**Impact of High PLS and Spent Electrolyte Temperatures on Degradation**

Agitated leaching systems, particularly those processing secondary sulphides, often operate above ambient temperature. As oxide deposits reach the end of their life and the processing of primary sulphides increases, an increasing prevalence of autoclave leaching is expected to be seen in this part of the world. The temperature of the PLS into SX is typically higher than that of atmospheric percolation leaching operations. Spent electrolyte is also typically at 40°C due to the resistive heating of the electrolyte during electrodeposition. The effect of elevated temperature manifests in two ways. The main mechanism of diluent loss is via evaporation. The rate of evaporation increases with increasing temperature, leading to increased diluent consumption per tonne of cathode produced. Elevated temperature also increases the rate of hydrolytic degradation of both extractant and diluent.

**Aqueous and Organic Entrainment**

One positive aspect of aqueous-in-organic entrainment—unique to Africa—is the transfer of cobalt from the PLS to the advance electrolyte. Copper EW typically requires 150 to 200 mg/L Co to be present in the electrolyte to assist with passivation of the anode and minimising lead contamination of the cathode. This is usually added as cobalt sulphate salt and comprises a significant operating cost (USD 3.50–4.00/t Cu) in North and South American operations. In Africa, the electrolyte typically contains some 2000 mg/L Co as a result of entrainment from the PLS (Sole et al., 2013) and no further cobalt addition is required. However, consideration of cobalt losses must therefore also drive decisions regarding the magnitude and destination of any bleed from the EW circuit.

Organic-in-aqueous entrainment represents the main source of extractant losses in these circuits. This is caused by crud formation and inefficient phase disengagement in settlers. Methods to recover organic include the installation of an after-settler, flotation columns, activated carbon columns or other coalescing systems. The use of coalescence packs in loaded organic tanks assists with removal of entrained PLS ahead of the strip circuit. It is critically important to minimise the organic contamination of the advance electrolyte (target <30 mg/L entrained organic). Not only does the presence of organic ruin the cathode appearance, but it has been responsible for several tankhouse fires in DRC in the past year.
CONCLUSIONS

As indicated in the discussions above, copper SX in Central Africa faces many challenges, but offers many significant opportunities in return. The current and planned expansions of existing operations and the proliferation of projects in various stages of development in the Copper Belt suggest that this region will become the dominant copper-producer of the world within the next decade. The rich, extensive and easily leachable ore bodies, coupled with low labour costs, offer many opportunities for large operators to move down the cost curve and become amongst the lowest cost producers. Hampering the rapid development of the industry are the ongoing (but improving) power supply situation, expensive and difficult logistics, lack of capacity in technical and management skills, corruption and unhelpful government regulation and legislation. The many unique issues presented by this region are promoting innovation and the development of new and interesting flowsheets, all of which enhance the use of SX technology. Provided that the political environment remains stable, the long-term outlook for the copper industry in this part of the world is extremely positive for operators, suppliers and engineering companies.

REFERENCES


