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Synopsis

The effect of mineralogy on the grindability was investigated using three copper ores – two sulphides and one oxide. The dominant copper minerals were identified by optical microscopy and mineral chemistry derived from SEM-EDS analysis. The sample designated sulphide 1 was borniterich, sulphide 2 ore was mainly chalcopyrite, and the oxide ore was predominantly malachite and minor azurite. The gangue minerals were identified using semi-qualitative XRD analysis. Sulphide 1 contained more than 80% (w/w) of quartz compared to about 70% in the other two ores. The Bond work indices were 13.8, 21.6, and 17.3 kWh/t for sulphide 1, sulphide 2, and oxide ore respectively. This suggested that the chalcopyrite-rich ore is the hardest, while the malachite-rich ore has intermediate hardness, and the bornite-rich ore is the softest. The brittleness indices of the ores were calculated using the chemical composition of the gangue, and a good correlation between brittleness indices and Bond work indices was observed, which highlights the importance of the gangue composition in determining the fracture behaviour of the ores. There is scope for further investigation into the relationship between ore mineralogy and comminution behaviour using other breakage characterization techniques.

Keywords

mineralogy, grindability, work index, relative work index, brittleness index, relative toughness index.

Introduction

Process mineralogy is concerned with the practical application of mineralogical knowledge of ores to guide the development and/or optimization of process flow sheets (Baum, 2014; Bradshaw, 2014; Henley, 1983; Lotter *et al.*, 2018; Whiteman, Lotter, and Amos, 2016). Ores are characterized by inhomogeneities such as flaws, joints, voids', and a variety of minerals dispersed as grains of various sizes with differing interrelationships with one another, *i.e.* texture (Chen, Xu, and Li, 2018; Estay and Chiang, 2013; Fan *et al.*, 2018; Hamdi, Stread, and Elmo, 2015; King, 2012; Zhao and Zhang, 2020; Zhou *et al.*, 2015). These discontinuities influence the hardness and fracture toughness of rocks and, in turn, their fracture behaviour (Gutierrez and Youn, 2015), serving as stress risers where both crack initiation and propagation begin (Chen, Xu, and Li, 2018; Griffith, 1920; Zhao *et al.*, 2016).

Mineralogical information is obtained from optical microscopy (mineral texture, mineral identification), scanning electron microscopy (SEM) (morphological characteristics), quantitative evaluation of materials by scanning electron microscopy (QEMSCAN) (degree of liberation of valuable minerals), mineral liberation analyser (MLA) (liberation data), and X-ray diffraction (XRD) (phase identification) (Anticoi *et al.*, 2018; Baum, 2014; Masindi, Foteinis, and Chatzisymeon, 2022; Shi and Zuo, 2014; Sipunga, 2015; Ye, Tang, and Xi, 2020; Liu *et al.*, 2017; Mariano, 2016; Palm *et al.*, 2010; Reichert *et al.*, 2015; Solomon *et al.*, 2011; Xiao *et al.*, 2012). In comminution, the mineralogical data can be related to the grindability parameters (*e.g.*, Bond work index) in order to develop the relationship between the ore mineralogy and grindability. The Bond work index (W_i) is the ore hardness parameter which represents the resistance to crushing or milling (Wills and Finch, 2016) and it is used to design comminution machines such as crushers and mills (Ipek and Goktepe, 2011; Jauregui, 1982; Todorovic *et al.*, 2017). Equation [1] denotes the energy required to reduce the parent particle size to the target progeny size. Additionally, W_i can serve as an input parameter for a geometallurgical model which predicts the grinding efficiency in the milling circuits (Mine, 2017; Mwanga, 2016).

$$E = W_i \left(\frac{10}{\sqrt{P_{80}}} - \frac{10}{\sqrt{F_{80}}} \right)$$
 [1]

where W_i is the Bond work index in kWh/t, and F_{80} and P_{80} are the 80% passing sizes (in μ m) for the feed and product respectively.

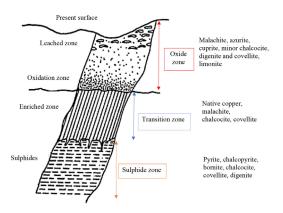


Figure 1—A typical copper deposit hosted by an arkosic sandstone, featuring three mineralization zones. The arkosic sandstone is dominated by quartz, K-feldspar, and minor phyllosilicates such as muscovite, chlorite, and biotite

The influence of ore texture has long been recognized as a critical factor in mineral processing (Bradshaw, 2014; Ghanei, 2020; Mine, 2017; Mwanga *et al.*, 2014; Schouwstra and Smit, 2011). For example, Mine (2017) developed the relationship between grindability and mineralogical data for the Aitik copper mine in Sweden. Ghanei (2020) investigated the relationship between ore texture and grindability in a variety of iron ores. Their work ignited the interest of the authors of this study to advance the understanding of the relationship between ore mineralogy and grindability using a copper deposit that contains multiple mineralized zones as a case study.

The deposit studied is located in the Pan African Damara Orogen, on the Northern Carbonate Platform section within the Mulden Group in the Otavi Mountainland, northern Namibia. The mineralization is hosted by arkosic sandstone with minor calcite (CaCO₃) and dolomite (CaMg(CO₃)₂). The deposit contains three mineralized zones (Kamona and Günzel, 2007). The uppermost zone consists of copper oxides, hydroxides, and carbonates. This is followed by a transition (mixed) zone and then the lowermost sulphide zone as shown in Figure 1. Copper mineralization in the oxide zone, from near the surface to about 70 m depth, is mainly composed of malachite (Cu₂CO₃(OH)₂), azurite (Cu₃(CO₃)₂(OH)₂), and cuprite (Cu₂O), with minor chalcocite (Cu₂S), digenite (Cu_{1-s}S), and covellite (CuS). The transition zone contains a combination of oxides (predominantly malachite) and sulphides (predominantly chalcocite, and bornite, with covellite (CuS) in minor quantities). The sulphide zone contains very fine intergranular disseminated grains of primary copper sulphide minerals (chalcopyrite (CuFeS₂) and bornite (Cu₅FeS₄)) and secondary copper sulphides such as chalcocite (Cu₂S), covellite (CuS), and digenite (Cu_{1-s}S).

Experimental

Three copper ore samples, designated sulphide 1, sulphide 2, and oxide, were obtained from the mine. About 200 kg of each ore type was homogenized using the coning and quartering method to obtain a representative sample weighing approximately 50 kg. The 50 kg samples were stage crushed to 100% passing 3350 μm (a suitable feed size for the laboratory ball mill) using a laboratory jaw crusher. The crushed 50 kg samples were rotary split to prepare 1 kg aliquots for the grindability experiments.

Elemental compositions of representative samples of the crushed feeds were determined using a bench-top X-ray fluorescence (XRF) instrument, model NEX CG supplied by Applied Rigaku Technologies from Austin, Texas, USA. This instrument uses a built-in calibration procedure referred to as the multi-channel analyser (MCA) calibration software. A representative sample was dry screened over 2360, 1700, 1180, 850, 600, 425, 300, 212, 150, 106,- and 75 µm aperture sieves to obtain the size distribution. After sieving, the size-by-size samples (-75, -300+75 and +300 μm) were prepared for chemical analysis by XRF, as well as for mineralogical examination with optical microscopy, X-ray diffractometry (XRD), and scanning electron microscopy (SEM). The samples for optical microscopy and SEM analysis were mounted in hot resin and then polished on Kemet waterproof silicon carbide grit papers using a 1 µm Kemet diamond slurry to obtain a reflecting surface. In preparation for SEM analysis, the samples were lightly carboncoated using the Quorum Q150T sputter coater (Advancedlab, Switzerland). SEM analysis was conducted using a JEOL JSM-IT300 scanning electron microscope coupled with the Thermo Scientific NS7 EDS software (Advancedlab, Switzerland). Analyses were done in high vacuum in the backscattered mode at an acceleration voltage of 15 kV, a probe current of 50 nA, and a working distance of 15 mm. The size-by-size samples were observed in both reflection and transmission modes for phase identification using an Olympus BX51 optical microscope. Sub-samples of the size-by-size samples were pulverized for semi-quantitative analysis of the gangue phases using a Bruker D8 Advance XRD instrument. The instrument uses the Diffracplus package with XRD Wizard version 2.9.0.22 for sample information entry, XDR Commander version 2.6.1.0 to run the samples, and EVA version 16.0.0.0 for data evaluation. The samples were run with a step size of 0.02 degrees 2Θ with a counting time of 1 second per step.

The Bond work indices (W_i) were experimentally determined following the Bond ball mill standard procedures outlined in Gupta and Yan (2006, pp. 73-75). For each sample type, size analysis was performed on the mill feed (100% passing 3.35 mm material) to determine the 80% passing size and percent passing the test sieve (106 µm). Following the iterative process of screening the mill discharge over the test sieve until three consecutive consistent readings for the undersize weight per revolution are obtained, the work index (in kWh/t) was then calculated using Equation [2]).

$$W_{i\text{Test}} = \frac{44.5}{D^{0.23} G_{bp}^{0.82} \left(\frac{10}{\sqrt{P_{80}}} - \frac{10}{\sqrt{F_{80}}}\right)}$$
[2]

where D is the test sieve aperture 106 μ m), and G_{bp} is the constant representing the net mass (in grams) of undersize product per revolution.

Milling experiments were conducted with 1 kg samples which were milled for 5, 10, 15, and 20 minutes using a laboratory ball mill with both the internal diameter and length equal to 305 mm. The milling conditions are listed in Table I. The slurry density was kept constant at 65% solids in all tests. The mill products were wetscreened using a root two series of sieves between 300 and 75 μm , dried at 105°C for 4 hours, and the size distributions generated for all three copper ores.

The powder filling (f_c), ball filling (J), and fraction of space (voids) between the balls, denoted as U, were calculated, and kept constant during the experiments as indicated in Table I. The powder filling (f_c) is the volume of the charged material expressed as a fraction of the mill volume and is calculated using Equation [3]. The porosity of the bed (ϕ) is theoretically equal to 40% with a bulk density of 4.65 g/cm³ for steel balls (Austin, Klimpel, and Luckie, 1984).

$$f_c = \frac{\text{(Mass of powder/Powder density)}}{\text{Mill volume}} \times \frac{1}{\text{(1-ϕ)}}$$
[3]

The ball filling (*J*), which defines the volume of balls at rest expressed as a fraction of the mill volume, is calculated using Equation [4].

$$J = \frac{\text{(Mass of balls/Ball density)}}{\text{Mill volume}} \times \frac{1}{(1-\phi)}$$
 [4]

The fraction of voids between the balls at rest is calculated using Equation [5], which relates f_c and J. An interstitial void volume (U) closer or equal to 100% is desired in the mill as it ensures efficient ball loading (Kiangi, 2011; Mulenga and Moys, 2014).

Table I											
Test conditions for milling experiments											
Mill critical speed (r/min)	Fraction of critical speed (%)	f _c (%)	J (%)	<i>U</i> (%)							
76.6	75	3.01	7.61	98.8							

$$U = \frac{f_c}{\phi \times I} \tag{5}$$

Results and discussion

Chemical analysis of the copper ores

The XRF elemental compositions for the bulk homogenized samples of the three copper ores are shown in Table II. As a way of validating the XRF assays, SEM-EDS was used to confirm the mineral phases and results are listed in Table A1 in Appendix A. The major difference for the metal values contained in the ores arises from the S and Ca contents. In general, the elemental compositions show that the two sulphide ores have a lower Ca grade and a higher S grade compared to the oxide ore. As already mentioned, the Ca and Mg are contained in the calcite (CaCO₃) and dolomite (CaMg(CO₃)₂) which are minor gangue minerals. Si, K, and Al are contained in the abundant gangue minerals, *i.e.* quartz (SiO₂), K-feldspar (KAlSi₃O₈), and muscovite (KAl₂(AlSi₃O₁₁)(OH)₂), which were identified using XRD (see Table III).

Table II

Elemental compositions for the feed size-by-size samples of the three copper ores, as determined by XRF

Ore	Particle size (μm)		XRF elemental composition (%)									
classification		Cu	Fe	S	Si	Ca	Mg	Al	K	Others (mainly O*)		
	+300	2.76	1.18	0.84	32.75	0.76	2.50	3.63	3.35	52.23		
Sulphide 1	-300+75	2.78	1.41	0.73	34.65	0.62	1.10	3.85	3.18	51.68		
	-75	3.73	1.74	0.82	32.61	0.61	0.67	5.04	2.86	51.92		
	+300	1.30	1.34	1.03	30.27	0.79	1.87	6.87	4.82	51.71		
Sulphide 2	-300+75	1.08	1.30	0.85	29.22	0.68	1.12	6.50	4.67	54.58		
	-75	2.01	2.71	1.24	28.53	0.81	1.35	7.68	4.71	50.96		
	+300	2.04	2.75	0.03	29.92	4.17	1.52	4.86	3.49	51.22		
Oxide	-300+75	1.19	3.02	0.03	27.33	2.36	1.12	5.29	3.61	56.05		
	-75	1.98	4.46	0.09	25.31	4.27	3.26	5.91	5.04	49.68		

^{*}Oxygen content can be observed in Table A1 in Appendix A from SEM-EDS

Table III						
XRD analysis of th	e gangue phases	in the three	e copper ores			
Ore classification	Particle size		Gangue minerals con			
	range (μm)	Quartz K-feldspars (microcline and orthoclase)		Muscovite	Total (%)	Others (%)
		SiO ₂	KAlSi ₃ O ₈	KAl ₂ (AlSi ₃ O ₁₁)(OH) ₂		
Sulphide 1	- 3350+300	82.1	5.1	-	87.2	12.8
	-300+75	86.0	6.0	-	92.0	8
	-75	87.5	12.5	-	100.0	0.0
Sulphide 2	-3350+300	65.2	4.9	-	70.1	29.9
	-300+75	77.6	4.7	-	82.3	17.7
	-75	65.9	7.2	8.2	81.3	18.7
Oxide	-3350+300	73.0	3.9	-	76.9	23.1
	-300+75	49.5	4.8	-	54.3	45.7
	-75	67.28	10.48	-	77.8	22.2

Table IV													
Physical properties of the copper and main gangue minerals in the ores													
Durananta	Domi	nant copper mine	erals		Major gangı	ie minerals							
Property	Bornite	Chalcopyrite	Malachite	Quartz	Microcline	Orthoclase	Muscovite						
Density (g/cm³)	5.09	4.1 - 4.3	3.6 - 4.0	2.65	2.56	2.56	2.82						
Mohs hardness	3.0	3.5 - 4	3.5	7.0	6.0 - 6.5	6.0	2 – 2.5						
Fracture	Conchoidal	Uneven	None	Conchoidal	Uneven	Uneven	Brittle						
Cleavage	Indiscernible	Poorly defined	No cleavage	Indistinct	Perfect on {001}	Perfect on {001}	Perfect on {001}						
Crystal structure	Orthorhombic	Tetragonal	Monoclinic	Trigonal- trapezohedral	Triclinic	Monoclinic	Monoclinic						

Mineralogical characterization

Semi-qualitative analysis of the main gangue minerals

Table III shows the semi-quantitative XRD results for the gangue minerals. The contents of copper minerals as well as iron sulphide minerals such as pyrite were below the detection limit for XRD, although they were detected by optical microscopy (see the following section). Considering the average composition, sulphide 1 contains 85% (w/w) quartz and 8% K-feldspars. Sulphide 2 constitutes 70% quartz, 8% K-feldspars, and 8% muscovite. The oxide ore contains 70% quartz (disregarding the –300+75µm fraction) and 7% K-feldspar minerals. Analysis of gangue minerals is essential because, as discussed later, the grindability of the ore is mainly dependent on gangue composition, gangue being the main constituent of the ore.

Optical mineralogy

The ore samples were not collected *in situ* but were received after having been pre-crushed. As such, their original textures were not easy to reconstruct. This section details the mineralogical assemblages in the crushed ores. Bornite particles (Figure 2a) exhibit deformational twinning; pyrite is elongated and fractured.

These features are associated with the comminution process. Pyrrhotite, a rare phase in these ores, occurs as small 0.05 mm grains (Figure 2a). Bornite, unlike chalcopyrite, exhibits the additional feature of grain-boundary sliding (Figure 2b). The association bornite+chalcopyrite+pyrite is the most common assemblage. Individual grains of chalcopyrite (cpy) and bornite are also present. In Figures 2d, 2e, and 2f a close association of chalcopyrite and pyrite with oxide phases, mainly malachite and minor azurite, can be observed. Based on the optical microscopy results, sulphide 1 has bornite as the dominant copper mineral, sulphide 2 is predominately chalcopyrite, and the oxide ore is mainly made up of malachite.

The physical properties of the main copper and gangue minerals in the ores are listed in Table IV. As regards the copper minerals, bornite is softer than chalcopyrite and malachite. Among the gangue minerals, quartz (the dominant phase, see Table III) is harder than the K-feldspar minerals (microcline and orthoclase). The phyllosilicate mineral muscovite is the softest of the main gangue minerals in the ores. The grindability of the ore is mainly affected by their main constituent (gangue minerals), as shown by the relationship between the gangue mineral composition and Bond work indices, which is discussed later.

Table V
Elemental analysis of the valuable and gangue mineral phases in the ore samples for the particle
size fraction -3350+300 um

Elements	ents SEM-EDS analysis (%)										
	Sulpl	hide 1	Sulpl	hide 2	Oxide ore						
	Phase 1	Phase 3	Phase 2	Phase 4	Phase 1	Phase 3					
Cu	27.18	-	13.17	-	59.44	-					
Fe	11.03	-	24.08	-	-	-					
S	22.71	-	38.42	-	-	-					
Si	12.82	42.52	4.78	30.50	1.82	31.26					
Ca	-	-	0.19	-	0.69	-					
Mg	-	-	0.24	-	-	0.14					
Al	0.26	0.92	2.76	9.80	-	10.22					
0	25.99	55.65	15.38	47.94	38.05	46.17					
K	-	0.92	0.97	11.42	-	11.92					
Na	-	-	-	- 0.34 -		0.28					
Total	99.99	100.01	99.99	.99 100.00 100.00		99.99					

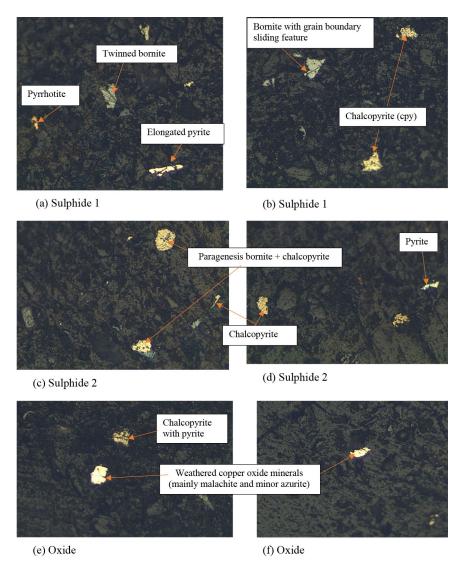


Figure 2 —Photomicrographs of the -75 μm fractions of the crushed ores. All images have a width of 0.4 mm

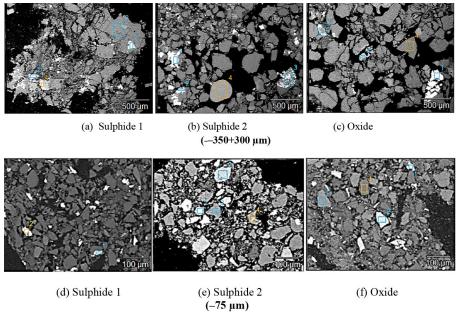


Figure 3—SEM BSE micrographs for –300+75 μm fraction (at 50× magnification) and –75 μm particles (at 200× magnification)

Scanning electron microscopy

SEM-EDS analysis was done on the three size fractions (-3350+300, -300+75, and $-75 \mu m$) to confirm the mineral phases in the ores. Due to similar observations for the three size fractions, only the backscattered electron (BSE) micrographs of the coarser $(-3350+300 \mu m)$ and fine $(-75 \mu m)$ size fractions are shown in Figure 3. Micrographs of the mid-size fraction are included in Appendix A, which also lists the SEM-EDS analyses of all the selected phases. In Figure 3, various phases can be seen. The heavy minerals appear as bright grains and the light (gangue) minerals dark. Table V lists the SEM-EDS analyses of some of the phases. The dominant elements in the heavy mineral phases are Cu, Fe, and S, which suggests sulphide minerals such as bornite, chalcopyrite, and pyrite. Cu and O are the dominant elements for the heavy mineral phase in the oxide ore, which indicates minerals such as malachite and azurite. The grey (abundant) phase is mainly composed of Si, Al, and K, which confirms the XRD results which indicated that quartz and feldspar are the dominant gangue minerals in the ores.

Grindability as a function of ore mineralogy

Size distribution of the feed

The size distributions for the jaw crusher products are shown in Figure 4. These samples were the feeds for the Bond ball mill work index tests which require the feed to be -3.35 mm. While the oxide (malachite) and sulphide 2 (chalcopyrite) ores had comparable size distributions, the sulphide 1 ore (bornite) was coarser. The effect of the discrepancies in the size distributions of the three ores on the analysis of results was eliminated through normalization of the experimental results (see Figures 5 and 6). The size analysis was done in duplicate for each ore.

Bond work index, brittleness, and toughness

The Bond work indices (W_i) for the three ores are listed in Table VI. Based on the standard classification of ore hardness in Table VII, sulphide 1 is of 'medium' hardness, the oxide ore is 'hard', and sulphide 2 is 'very hard'. The relative work indices (W_r) were also calculated with respect to to the ore with the lowest work index (i.e., work index of the ore divided by the work index of sulphide 1). The

distribution of the copper mineral grains (bornite, chalcopyrite, malachite, and azurite) within the host rocks introduces structural defects which serve as sites for crack initiation and propagation upon the application of an external load on the particles during comminution (Charikinya, Bradshaw, and Becker, 2015; Chen, Xu, and Li, 2018; Fan et al., 2018; Germanovich et al., 1994; Ghorbani et al., 2013; Griffith, 1920; Shi, Zuo, and Manlapig, 2013; Viljoen et al., 2015; Zhao et al., 2016). The difference in the Wi values for the three copper ores can thus be attributed to mineral texture. However, the elemental compositions (Table II) indicate that the contents of copper minerals such as bornite, chalcopyrite and malachite in the three ores are below 7% (w/w) and their volume fractions are lower than 4% (v/v). Therefore, the Bond work indices of the ores are mainly dependent on the fracturing behaviour of the gangue (abundant) minerals (Buntoro et al., 2018). The lowest W_i , for sulphide 1, could be attributed to the conchoidal fracturing of quartz (see Table IV), since this ore has the highest quartz content Table III). Based on Mohs scale of hardness, quartz is the hardest of all the minerals identified in the ores. Kim et al. (2022) reported that hardness and brittleness are positively correlated with quartz content. The XRD results in Table III, considering the -75 and $-3350+300 \mu m$ size fractions (the results for the $-300+75 \mu m$ fraction were disregarded because of the anomaly observed in the quartz content for the oxide ore) showed that sulphide 1, sulphide 2, and oxide contained 85%, 66%, and 70% (w/w) quartz respectively. It can be inferred that the brittleness is highest in sulphide 1, intermediate in the oxide ore, and lowest in sulphide 2. The trend agrees with the W_i values, suggesting that sulphide 2 requires more energy to grind (highest W_i , lowest brittleness), followed by the oxide ore (intermediate W_i and intermediate brittleness), and finally sulphide 1 requires less comminution energy (lowest W_i and highest brittleness).

In the effort to establish the relationship between the chemical composition and W_i , the brittleness indices for the ores were determined. The brittleness index (β_i) , which is used in fracture mechanics, is dependent on various factors such as composition, texture, porosity, and the applied external load to effect fragmentation (Ye, Tang, and Xi, 2020). Various methods (Fatah et al., 2016; Ye, Tang, and Xi, 2020) are currently applied to calculate

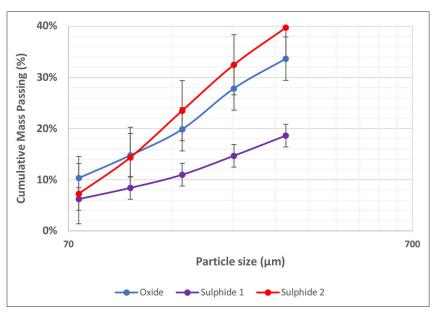


Figure 4—Size distributions for the mill feeds

Table VI											
Bond work indices, brittleness indices, and relative toughness values of copper ores											
Ore classification	W _i (kWh/t)	Relative Bond work, W _r	Brittleness index, β_i	Relative toughness, κ _r							
Sulphide 1 (bornite)	13.82	1.00	17.68	1.00							
Sulphide 2 (chalcopyrite)	21.61	1.56	11.19	1.58							
Oxide (malachite)	17.26	1.26	13.06	1.35							

Table VII	Table VII										
'-	Typical classification of ore hardness in terms of BBWI (Napier-Munn <i>et al.</i> , 2005)										
Wi (kWh/t)	7-9	10-14	15-20	>20							
Classification	Soft	Medium	Hard	Very hard							

the brittleness index, and these can be grouped into two categories: (i) brittleness indices based on rock mechanical properties using parameters such as stress, strain, Poisson's ratio, and Young's modulus (Guo, Chapman, and Li, 2012; Jiang *et al.*, 2020; Li *et al.*, 2020; Meng, Wong, and Zhou, 2021; Mews, Alhubail, and Barati, 2019; Ozkahraman, 2010; Swain and Rao, 2009), and (ii) brittleness indices derived from mineral content (Buntoro *et al.*, 2018; Fatah *et al.*, 2016; Ye, Tang, and Xi, 2020). Based on the mechanical properties, β_i is obtained by dividing the compressive strength by tensile strength (Meng, Wong, and Zhou, 2021). In the later method the brittleness index is the weight ratio of the brittle minerals, such as quartz and feldspar, to the total mineral content (Fatah *et al.*, 2016; Kang *et al.*, 2020; Ye, Tang, and Xi, 2020). While the β_i values from the two methods are not numerically equal, they give a relative indication of the brittleness of rocks.

The later method of calculating the brittleness index, but based on the chemical (elemental) composition rather than the mineral composition, was considered in this study. β_i was defined using Equation [6] as the ratio of the acidic elements Si and Al, which contribute to the brittleness of the ore (Buntoro et al., 2018) to the basic elements (K, Ca, and Mg). This is a reciprocal of the basicity modulus in chemistry (Humad, Habermehl-Cwirzen, and Cwirzen, 2019). Pyrite is reported to contribute to the brittleness (Kang et al., 2020), but since it is a minor gangue mineral relative to the silicates, it was not considered for brittleness indexation. The calculated brittleness indices for the three copper ores are listed in Table VI. In agreement with the W_i values in reverse proportionality, it is observed that sulphide 1 (with the smallest W_i) has the highest β_i , sulphide 2 (with the highest W_i) the smallest β_i , and the oxide ore (intermediate W_i) the intermediate β_i . The trend of the calculated β_i agrees with the results reported by Kim et al. (2022).

$$\beta_i = \frac{\text{wt\%Si+wt\%Al}}{\text{wt\%K+wt\%Ca+wt\%Mg}}$$
 [6]

The relative toughness (κ_r) of the ores were then calculated relative to the most brittle (sulphide 1 with bornite as a dominant copper mineral) as follows:

$$\kappa_{\rm r} = \frac{\beta_{\rm bornite \, ore}}{\beta_{\rm i}} \tag{7}$$

The similarity of the relative toughness values(κ_r) and the corresponding relative Bond work indices(W_r) in Table VI shows

the importance of the chemical composition of the gangue minerals in characterization of the fracture behaviour of ores during comminution.

The three copper ores were mined from different mineralization zones within the same ore deposit and thus the experimental W_i values of the distinct ores could be useful for the development of the geometallurgical model for that deposit (Mine, 2017; Mwanga et al., 2014). However, to account for the inherent variations in the orebody, many samples are needed to develop a reliable geometallurgical model. The procedure used here to determine W_i is time-consuming and thus alternative comminution tests that provide information on the hardness of the ores are used to generate data for geometallurgical modelling. These tests include, among many others, the uniaxial compressive strength (UCS) test, which gives the strength of the rock under compressive stress (Akinbinu, 2017; Ozkahraman, 2010), and the drop weight tests using single particles of narrow size fractions to determine the 10% passing sizes (commonly referred to as parameter *t10*, which allows the determination of the parameters (denoted as *A* and *b*, in the model shown in Equation [8]), the product of which $(A \times b)$ indicates the ore's resistance to impact breakage (Ballantyne, Bonfils, and Powell, 2017; Genç, Ergün, and Benzer, 2004; Nghipulile et al., 2021; Nghipulile, 2019; Tavares, 2007).

$$t_{10} = A \times (1 - e^{-bE_{cs}})$$
 [8]

where A and b are material-specific impact breakage parameters.

Fitting feed and product size distributions to the Weibull function

The particle size distributions for the feed and mill products after grinding the 1 kg samples for 5, 10, 15, and 20 minutes were fitted to the Weibull distribution, which is described by the following equation:

$$P(x) = 1 - e^{-\left(\frac{x}{x_c}\right)^m} \tag{9}$$

where P(x) is the cumulative percentage passing (%) corresponding to the fragment size $(1-\frac{1}{\varepsilon})$; the term x_c , called the modular particle size, is the particle size corresponding to cumulative percentage passing $(1-\frac{1}{\varepsilon})$, and m is a fitting parameter, which indicates the width of the distribution.

Table VII-I												
Averaged values for width of cumulative mass distribution (m) and the standard errors												
Ore type	Average m	Standard deviation (σ)	Standard error (SE)									
Sulphide 1 (bornite ore)	1.35	0.222	0.0996									
Sulphide 2 (chalcopyrite ore)	1.46	0.189	0.0849									
Oxide ore	1.14	0.075	0.0334									

The cumulative mass distributions, both for the experimental data (shown as points) and the modelled Weibull distributions are presented in Appendix B. The fitting parameters in Equation [8] were determined through iteration by finding the minimum root mean square error (RMSE) between the experimental and modelled data. The Weibull distribution fits well to the experimental data, as shown in the plots in Appendix B. The widths of the distributions (parameter *m*) for the three ores are shown in Table VIII. It can be seen that the widths of distribution for the sulphide ores are greater (averaging 1.4) that for the oxide ore (1.1). The relationship between the *m* values and mineral composition has not yet been investigated. A meaningful relationship will only be established as more data become available.

As shown in Figure 5, the x_c values were normalized to the feed modular particle sizes to allow better comparison of the three ores through elimination of the variation in the feed size distributions. The normalized modular particle size is a dimensionless quantity which is denoted x'_c . To understand the modular size, say after 5 minutes of grinding, the normalized x_c values are 0.41, 0.30. and 0.21 for sulphide 2, oxide, and sulphide 1 respectively. The modular particle size (x_c) represents the particle size corresponding to 63% (w/w) passing. It is observed that the sulphide 2 ore (chalcopyrite) has the highest 63% passing sizes for any milling time ranging between 0 and 20 minutes. This suggests a low size reduction ratio (lowest β_i) for this ore, *i.e.* more energy is required to effect the required size reduction, as already indicated by the highest W_i . In line with the trend for the relative work indices (W_r) and relative toughness index (κ_r), the oxide ore has intermediate modular

particle size, and sulphide 1 the smallest, suggesting a high size reduction ratio is achievable among the three ores assuming the same input energy.

Conclusions

The Bond ball mill work indices suggest that the chalcopyrite ore $(W_i \text{ of } 22 \text{ kWh/t})$ is the hardest and bornite ore $(W_i \text{ of } 14 \text{ kWh/t})$ the softest, while the malachite ore (W_i of 17 kWh/t) has an intermediate resistance to breakage. The W_i values were inversely proportional to the quartz content (which increases with the brittleness) in the -3350+300 μm and -75 μm size fractions. Sulphide 1 had an average quartz content of 85% (w/w), sulphide 2 had 66%, and the oxide ore had 70%.

Brittleness and relative toughness indices, based on the chemical composition of the gangue minerals in the ores, were calculated. A good correlation between the relative toughness and the corresponding relative Bond work indices highlighted the importance of the chemical composition of the gangue, the dominant component of the ore, when characterizing the comminution behaviour.

The information from this study, especially the work indices, can be useful when designing a milling circuit to process the ores, as the specific comminution energy required to reduce the parent feed size to the target product size can be estimated using W_i . It is recommended that the relationship between the mineralogy and comminution behaviour should also be investigated using other breakage characterization methods, such as the uniaxial compressive strength test, Brazilian tensile strength test, and drop weight test.

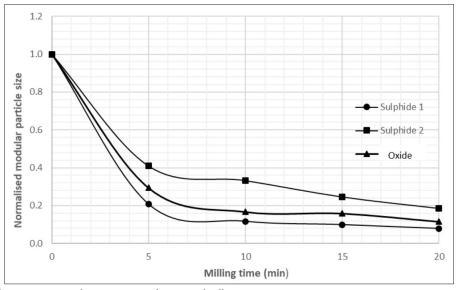


Figure 5—The Weibull distribution parameters for copper ores as functions of milling time

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Appendix A

Table A1													
Elemental compositions for the feed size-by-size samples of the three copper ores, as determined by SEM													
Ore classification	Particle size (µm)		SEM EDS elemental composition (%)										
		Cu	Fe	S	0	Si	Ca	Mg	Al	K			
Sulphide 1	+300	2.55	0.33	0.79	54.49	36.81	0.61	0.59	2.79	1.04			
	-300+75	1.94	0.54	1.21	53.24	37.05	0.87	0.47	2.62	0.59			
	-75	1.89	1.12	1.23	50.93	36.05	0.69	0.77	4.60	2.12			
Sulphide 2	+300	1.64	1.11	1.83	51.16	34.97	0.52	0.37	5.66	2.74			
	-300+75	0.43	0.90	1.30	54.00	33.90	0.49	0.43	5.26	2.44			
	-75	1.17	1.74	2.45	49.53	32.33	0.51	0.70	7.35	2.79			
Oxide	+300	1.63	0.51	0.13	51.96	38.18	1.53	0.49	3.34	1.16			
	-300+75	1.32	2.52	1.40	49.83	33.61	1.75	0.50	4.94	2.17			
	-75	0.93	2.97	0.45	50.88	27.08	2.93	0.87	7.46	3.15			

 $\label{eq:analysis} \emph{Table A2}$ Elemental analysis of the valuable and gangue mineral phases in the ore samples for the particle range $-3350+300~\mu m$

		SEM EDS analysis (%)												
Elamanta		Su	lphide 1			Sul	phide 2		Oxide					
Elements	Bright phases			Grey phases	В	right phas	es	Grey phases	Bright	nt phases Grey p		phases		
	Phase 1	Phase 2	Phase 4	Phase 3	Phase 1	Phase 2	Phase 3	Phase 4	Phase 1	Phase 2	Phase 3	Phase 4		
Cu	27.18	58.96	34.31	-	56.43	13.17	8.58	-	59.44	-	-	-		
Fe	11.03	-	-	-	-	24.08	5.44	-	-	-	-	-		
S	22.71	19.33	11.44	-	19.11	38.42	10.05	-	-	-	-	-		
Si	12.82	7.63	18.88	42.52	12.11	4.78	33.90	30.50	1.82	33.81	31.26	46.96		
Ca	-	-	-	-	-	0.19	-	-	0.69	1.31	-	-		
Mg	-	0.67	-	-	-	0.24	-	-	-	0.22	0.14	-		
Al	0.26	0.72	5.17	0.92	-	2.76	0.12	9.80	-	0.95	10.22	0.17		
0	25.99	12.69	25.93	55.65	12.35	15.38	41.91	47.94	38.05	52.43	46.17	52.88		
K	-	-	3.50	0.92		0.97	-	11.42	-	0.29	11.92	-		
Na	-	-	-	-	-	-	-	0.34	-	-	0.28	-		
P	-	-	0.54	-	-	-	-	-		-	0.94	-		
Ti	-	-	0.22	-	-	-	-	-	-	10.06	-	-		
Total	99.99	100.00	99.99	100.01	100.00	99.99	100.00	100.00	100.00	100.01	99.99	100.01		

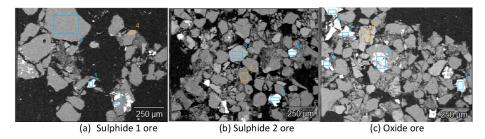


Figure A1—SEM micrographs of -300+75 μm particles

 ${\it Table~A3}$ Elemental analysis of the valuable and gangue mineral phases in the ore samples for the -300+75 \$\mu m\$ fraction

							SEM EDS	S analysis	(%)					
		Sulpl	hide 1		Sulphide 2				Oxide					
Elements	Bright phases Grey phases		Bright	phases	Grey	phases		F	Bright phase	es		Grey phases		
	Phase 1	Phase 2	Phase 3	Phase 4	Phase 1	Phase 2	Phase 3	Phase 4	Phase 1	Phase 2	Phase 3	Phase 4	Phase 5	Phase 5
Cu	-	64.97	-	-	4.70	3.32	-	-	54.93	24.65	-	-	-	-
Fe	43.67	-	-	-	39.06	31.80	-	-	1.06	0.49	42.84	29.06	39.69	-
S	55.33	20.55	-	-	54.74	44.63	-	-	-	12.91	56.89	0.94	52.47	-
Si	0.65	6.78	46.91	31.57	1.50	4.40	29.16	42.86	2.48	2.11	0.27	20.43	2.59	31.40
Ca	-	-	-	-	-	-	-	-	-	-	-	2.14	-	-
Mg	-	-	-	-	-	-	-	-	-	-	-	3.58	0.33	-
Al	0.35	-	-	9.98	-	0.55	9.51	-	-	3.93	-	5.10	2.27	10.16
0	-	7.70	53.09	46.39	-	15.02	50.49	57.14	41.10	48.45	-	36.67	-	47.14
K	-	-	-	11.63	-	0.28	10.43	-	-	6.57	-	0.69	0.59	11.11
Na	-	-	-	0.43	-	-	0.41	-	-	0.89	-	1.39	-	0.19
P									0.43	-	-	-	-	-
Ti											<u> </u>		2.06	-
Total	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00

Table A4Elemental analysis of the valuable and gangue mineral phases in the ore samples for the $-75 \, \mu m$ fraction

Elements	SEM EDS analysis (%)									
	Sulphide 1 ore		Sulphide 2 ore				Oxide			
	Bright phases		Bright phases		Grey phases		Bright phases		Grey phases	
	Phase 1	Phase 2	Phase 1	Phase 2	Phase 3	Phase 4	Phase 1	Phase 2	Phase 3	Phase 4
Cu	51.33	71.3	49.64	0.09	-	-	59.8	-	-	-
Fe	8.98	-	6.24	3.35	-	-	-	27.20	-	-
S	25.35	23.24	24.33	-	-	-	21.12	0.88	-	-
Si	2.42	2.25	4.09	23.73	44.92	39.90	5.52	18.72	29.42	43.60
Ca	0.21	-	0.12	-	-	-	0.47	1.94	-	-
Mg	0.25	-	0.15	2.55	-	-	0.57	3.40	-	-
Al	-	0.19	3.88	13.23	0.07	3.17	1.26	4.76	9.64	0.11
О	9.79	3.01	11.32	48.90	55.01	53.44	10.97	40.90	49.94	56.29
K	-	-	0.23	7.31	-	3.50	0.29	0.75	10.60	-
Na	-	-	-	-	-	-	-	1.45	0.19	-
P	0.23	-	-	-	-	-	-	-	-	-
Ti	-	-	-	0.75	-	-	-	-	0.21	-
Br	1.17	-	-		-	-	-	-	-	-
Cr	-	-	-	0.10	-	-	-	-	-	-
Total	99.73	99.99	100.00	100.01	100.00	100.01	100.00	100.00	100.00	100.00

Appendix B: Cumulative mass distributions fitted to Weibull distribution

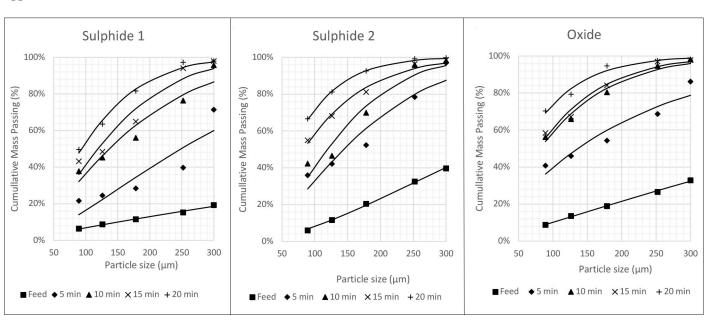


Figure B1—Experimental size distributions fitted to Weibull distribution