

by P.M. Mashaba<sup>1</sup> and S.O. Bada<sup>1</sup>

#### Affiliation:

<sup>1</sup>DSI-NRF SARCHI Clean Coal Technology Research Group, School of Chemical & Metallurgy, Faculty of Engineering and the Built Environment, University of the Witwatersrand, South Africa.

Correspondence to: P.M. Mashaba

Email: poitahmashaba@gmail.com

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ORCID: S.O. Bada https://orcid.org/0000-0002-1079-3492

P.M. Mashaba https://orcid.org/0000-0002-5454-9301

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# **Synopsis**

An advanced high internal phase water-in-oil (HIP W/O) emulsion binder and flotation with an internal dispersed water phase of 95 vol% was used to beneficiate South African weathered coal from discard dumps. Five coal samples with calorific values ranged from 8.48 MJ/kg to 20.94 MJ/kg were utilized. The performance of the HIP W/O emulsion was compared to that of kerosene in terms of the physicochemical properties of the clean coal products. A 5 kg/t emulsion binder addition produced a high-quality coal concentrate with 11.76% ash and a heating value of 28.61 MJ/kg from a sample containing 29.34% ash. Using the same dose and coal sample, kerosene resulted in a clean coal concentrate with 17.47% ash and a calorific value of 26.42 MJ/kg. Further samples were also beneficiated using 5 kg/t of W/O emulsion. The highest increase in calorific value was achieved from the beneficiation of a sample containing 48.71% ash, from 8.48 MJ/kg to 18.16 MJ/kg. It has been established that emulsion binders can upgrade coal samples dumped many decades ago.

## **Keywords**

calorific value, discard coal, rapid beneficiation, W/O emulsion technique, weathered coal.

# Introduction

Run of mine (ROM) coal is processed to separate the valuable carbonaceous component from the inorganic minerals. However, gravity beneficiation processes are restricted to certain particle size fractions and are ineffective for coals of less than 150 µm, generated through mechanized mining (Horsfall, Zitron, and de Korte., 1986; Liao *et al.*, 2017; Barma, 2018). In South Africa, about 1.5 Gt of discard coal have been produced, and a further 60 Mt are generated annually (Department of Energy, 2001). The South African Coal Roadmap indicated that fines and ultra-fines account for 11% of the annual coal discard (Fossil Fuel Foundation, 2011). Almost 100% of these fines are regarded as waste and stored in tailings dams (Bada, Falcon, and Bergmann, 2016).

Exposure of the waste coal to the atmosphere causes the production of hydrophilic oxygenated groups such as the hydroxyl (C–OH), carbonyl (–C=O), and carboxyl (O=C–OH) groups on the surface of the coal. As a result, the flotation of the coal using traditional reagents like kerosene or diesel becomes less effective with the application of physico-chemical processes such as froth flotation and selective agglomeration, and greater amounts of these organic liquids are required for successful beneficiation (Xia, Yang, and Liang, 2013; Li et al., 2018; Wang et al., 2018). Previous research has shown that for effective separation, an oil dosage of 10 to 20 weight per cent (wt%) is required to form a continuous oil layer on the coal surface and to fill the void spaces within the agglomerates in the traditional agglomeration process (Garcia, Vega, and Martinez-Tarazona, 1995; Dickinson, Jiang, and Galvin, 2015; Netten, Moreno-Atanasio, and Galvin, 2016). With the high internal phase water-in-oil (HIP W/O) emulsion technique, most of the void spaces between the coal particles are filled with dispersed water droplets, rather than oil as in traditional agglomeration. Using this approach, it has been found that the same quality product as that obtained by traditional agglomeration and flotation can be achieved with on-tenth the amount of organic liquid (Netten, Moreno-Atanasio, and Galvin, 2016; Netten and Galvin, 2018; and Lu et al., 2019).

The present investigation, the HIP W/O emulsion binder was used to beneficiate some weathered coals of 20 to 30 years old from various dumps. The use of HIP W/O emulsion binder with flotation is very similar to the selective agglomeration process. The binder acts as a hydrophilic and hydrophobic reagent

for selecting, wetting, and bonding hydrophobic and hydrophilic carbonaceous particles, thereby leaving the inorganic minerals unwetted (Netten, Borrow, and Galvin et al., 2017; Netten and Galvin, 2018; Lu et al., 2019). The viscosity of the emulsion binder is also known to influence the recovery rate in this process. Traditional reagents such as pure kerosene have a low viscosity of 0.0016 Pa, whereas HIP emulsion, depending on its composition, has a viscosity of around 100 Pa (Galvin et al., 2001). A high-viscosity solution or binder is expected to adhere to a particle after a collision with greater effectiveness than a low-viscosity binder. In addition, based on the composition of the emulsion and its stability, most of the void spaces between the coal particles are filled with dispersed water droplets rather than oil as in traditional agglomeration (Garcia, Vega, and Martinez-Tarazona, 1995; Dickinson, Jiang, and Galvin, 2015; Netten, Moreno-Atanasio, and Galvin, 2016).

According to the literature, over 80% of the volume of the dispersed phase comprises NaCl and water, leading to the reduction in the amount of organic liquids used in the preparation of the HIP W/O emulsion binder (Netten, Borrow, and Galvin, 2017; Netten and Galvin, 2018; Lu et al., 2019). Using an emulsion binder with 95% by volume of the dispersed phase, Netten and Galvin (2018) beneficiated coal feed that had been stored in water to limit surface oxidation. A clean coal product with an ash content of 15 to 17% and combustible recovery of 70 to 80% was obtained from feed coal with an ash content of 42%. Lu et al. (2019) beneficiated a laboratory-oxidized coal (-74 +38 µm) using an emulsion binder composed of 85% internal dispersed water phase by volume. The new collector improved the flotation of the oxidized coal. This highlighted the need for further study on the use of the HIP W/O emulsion binder can be used to beneficiate weathered coal fines from discard dumps or slurry ponds. In this study, five coal fines samples aged between two and three decades with different degrees of oxidation were studied by flotation. The performance of the HIP W/O emulsion with an internal dispersed water phase of 95% by volume was compared with that of kerosene, and the quality of the clean coal products recovered from the five coals were compared.

# **Experimental**

## Coal preparation and characterization

Five samples with various ash contents were collected from mines in the Witbank Basin (Figure 1). Greyish samples 2 and 4 were

collected from the outside of a stockpile where they have been severely exposed to atmospheric conditions. Sample 1, 3, and 5 were taken from within a stockpile. Total sulphur and total carbon analyses were performed according to ASTM D 4239-14:2015 using a LECO CHN 628 instrument with an add-on 628 S module. Proximate analysis, the calorific value (heating value), and FTIR analysis were also conducted on the samples in accordance with ASTM D-5142, ASTM D5865-04, and ASTM E168, respectively.

#### Reagents

Figure 2 shows the reagents used to beneficiate the coal samples. The emulsion binder comprised 5% organic liquid by volume, with the remainder being the dispersed phase (Figure 2a). Kerosene was used at 100% by volume (Figure 2b).

## Emulsion binder preparation

The organic liquids, kerosene, and Span 80, also known as Sorbitan monooleate (SMO), were mixed in equal proportions (7.5 ml each) in a standard domestic food mixer. An aqueous NaCl dispersion phase solution was formed by dissolving 8.55 g of NaCl in 285 ml of tap water. Subsequently, the NaCl solution was added while mixing to the organic liquid mixture in small increments of 7.5 ml every 15 seconds to prevent deterioration of the emulsion binder (Netten, Moreno-Atanasio, and Galvin, 2016). A viscous and opaque emulsion binder was produced (Figure 2a).

## Beneficiation test

Figure 3 illustrates the laboratory beneficiation flow sheet. A mechanical flotation cell was used to beneficiate the coal samples. The clean coal products from the rougher and scavenger stages were subjected to two cleaning stages. The coal slurry (25 wt% coal) was agitated in a 1.5 L flotation cell for two minutes at a speed of 1 000 r/min. The same process conditions were applied for two minutes in the scavenger, cleaner, and re-cleaner stages for both reagents (emulsion binder and kerosene).

The coal concentrate was scraped off after 30 seconds of aeration. The products obtained, including the clean concentrate and tailings (inorganic minerals), were oven-dried for 15 hours at a temperature of 50°C. The ash content, combustible recovery, yield and ash rejection of the dried products were then determined using Equations [1-4].

$$Ash~(\%)~=\frac{\textit{Residue remaining after burning (g)}}{\textit{Initial weight of carbonaceous ultra-fines (g)}}~\times 100~~[1]$$



Figure 1—Different discard coal samples taken



Figure 2—HIP W/O emulsion binder (a) and kerosene (b)

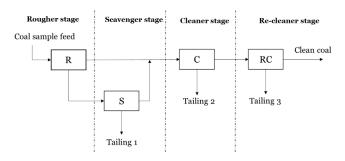


Figure 3—W/O emulsion technique beneficiation stages R - rougher cell, S - scavenger cell, C - cleaner, RC - re-cleaner cell

Yield (%) = 
$$\frac{\text{Weight of floated carbonaceous product}}{\text{Total feed weight}} \times 100$$
 [2]

$$Ash\,rejection\,(\%)\,=\,100\,-\,(\textit{Cumulative yield}\,\times\,(\frac{\textit{Cumulative ash}\,(\%)}{\textit{Feed ash}\,(\%)}))\ \ \, \left[4\right]$$

where  $M_c$  is the mass of the floated clean coal (g) (dried basis),  $M_F$  is the mass of the dry feed coal sample (g) (dried basis),  $A_C$  is the ash content of the dry floats (%), and  $A_F$  is the ash content of the dry coal feed (%).

# Reagent dosages

Table I presents the reagent doses. The doses were determined using Equation [5-7].

Reagents dosages 
$$(kg/ton) = \frac{mass\ of\ emulsion\ or\ kerosene}{mass\ of\ dry\ coal}$$
 [5]

Organic liquids within the emulsion (kg/ton)= binder  $(kg/ton) \times (5 \text{ vol}\%)$  [6]

Organic liquid within kerosene 
$$(kg/ton)$$
= kerosene  $(kg/ton) \times (100 \text{ vol}\%)$ 

# Results and discussion

#### Coal characteristics

The particle size distributions of the coal samples are presented in Figure 4. The percentage coal passing 150  $\mu$ m was 96% for sample 1, and 84%, 93%, 78% and 92% for samples 2, 3, 4 and 5, respectively. The difference in the size distribution curves can be attributed to the difference in hardness and friability of the coals (Isaac, 2019). Sample 2 had the highest ash content of 48.71%, followed by sample 3 (41.71%), and sample 4 with the lowest ash

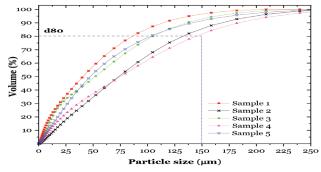


Figure 4—The particle size distribution of the coal samples

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# Results of KC tests on CFA sample

HIP emulsion and kerosene			Organic liquids used			
Doses	g	kg/t	HIP W/O emulsion		Kerosene	
			vol%	kg/t	vol%	kg/ton
1 2 3	1.5 2.0 2.5	5 7 9	5	0.25 0.35 0.45	100	5 7 9

Table II

The physico-chemical properties of the coal feeds

Coal sample	Proximate analysis (air-dry basis)			Total sulphur and carbon analysis (dry basis)		Calorific value (MJ/kg)	
	Ash content (%)	Fixed carbon (%)	Volatile matter (%)	Volatile matter (%)	Total carbon (%)	Total sulphur (%)	
1 2 3 4 5	29.34 48.71 41.71 28.63 33.32	43.29 17.79 35.58 43.83 40.68	23.40 28.69 20.79 22.88 22.63	3.02 5.80 7.36 5.71 5.51	54.50 24.9 43.7 55.95 51.45	0.87 3.04 1.29 1.00 0.43	20.9 8.48 16.29 20.61 18.55

content of 28.63% (Table II). The as-received coal samples are of low quality (grade D III) as indicated by the calorific values which range from 8 to 21 MJ/kg. Based on the proximate analysis results, the samples had fixed carbon contents ranging between 17.79% and 43.83%. Sample 2 had the highest total suphur content of 3.04%.

# FTIR spectra for the five coal samples, W/O emulsion, and kerosene

The FTIR spectra of the five coal samples are shown in Figure 5. This was undertaken to determine the degree of weathering of the coal surfaces. Coal oxidation is denoted by the intensity of the oxygenated functional groups such as the hydroxyl (O-H) and carbonyl (C=O) (Wang *et al.*, 2012). A distinct and intense peak of O-H groups is present for sample 2, suggesting that the coal is highly weathered. Although samples 4 and 5 show evidence of intense C=O groups, sample 3 contained the second-highest amount of the O-H groups.

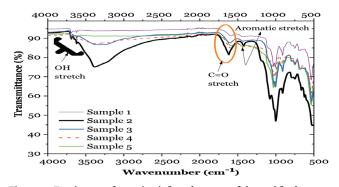


Figure 5—Fourier transformation infrared spectra of the coal feeds

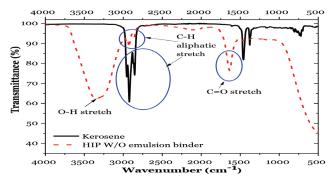


Figure 6—Spectra (4000 – 500 cm-1) for the reagents used to beneficiate the coal feeds

FTIR was also used to characterize the reagents to determine the functional group compositions (Figure 6). As indicated, the presence of O-H stretch and C=O stretch on the surface of HIP W/O emulsion binder are evident whereas none was observed on the surface of kerosene. The only groups present on the surface of kerosene are aliphatic groups 2837-2967 cm<sup>-1</sup>. This group is also available on the emulsion binder, suggesting that the emulsion is an emulsifier, *i.e.* comprises both oxygenated and aliphatic groups and can therefore be expected to bind to weathered coal.

# Washability, ash rejection, and physico-chemical properties of coal sample 1

# Influence of reagent dosage

The clean coal product yield, ash content and combustible recovery increase with increasing in kerosene and W/O emulsion binder dosage (Figures 7 and Figure 8). The observed trend corresponds to the findings of Netten, Moreno-Atanasio, and Galvin. (2016) and Lu *et al.* (2019). Netten and Galvin (2018) also showed that product ash content increased with combustible recovery due to an increase in emulsion binder dosage. According to Netten and Galvin (2018) the increase might be because of the binder sufficiency, which enhanced the floatability and pulling of semi-hydrophobic carbonaceous particles to the clean coal product. This therefore leads to a lower-grade carbonaceous product.

In this study, clean coal products with low ash contents of 17.47% and 11.76% ash were obtained using 5 kg/t kerosene and HIP W/O emulsion binder, respectively. A yield of 12.50% and combustible recovery of 15.61% was obtained using 5 kg/t emulsion (Figure 7). This was lower than the clean coal yield of 18.47% and recovery of 21.59% using 5 kg/t kerosene (Figure 8). Nevertheless, an abundance of pure kerosene was required to enhance the floatability of the carbonaceous particles the coal discard. This

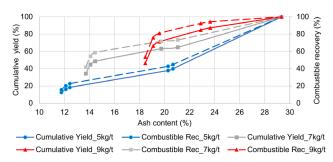


Figure 7—Effect of W/O emulsion dosages on clean coal 1 yield and ash content

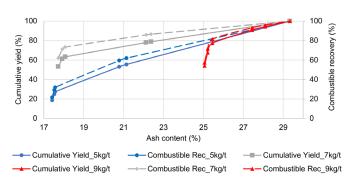


Figure 8—Impact of kerosene concentration on clean coal 1 product yield and ash content

agrees with the literature that for an effective separation using pure oil (diesel, kerosene *etc*), a higher oil dosage is essential (Garcia, Vega, and Matnines-Tarazona, 1995; Dickinson, Jiang, and Galvin, 2015; Netten, Moreno-Atanasio, and Galvin, 2016). The low recovery for sample 1 may be due of the presence of carbonyl groups (C=O), which enhance the potential of the air bubbles to repel the carbonaceous particles in the coal and lower the attachment rate (Sarikaya and Ozbayoglu, 1990; Li *et al.*, 2018). In comparison, the W/O HIP emulsion binder required only 0.25 kg/t organic oil, while 5 kg/t pure kerosene was required to beneficiate the same coal.

## Selectivity

Selectivity is the ability to differentiate hydrophobic and hydrophilic carbonaceous particles from inorganic mineral matter prior to advanced agglomeration and froth flotation (Netten and Galvin, 2018; Lu *et al.*, 2019). The extent to which the W/O emulsion binder and kerosene were able to select clean coal from the inorganic minerals in the coal sample was evaluated using ash rejection percentage (Equation [4]). As seen in Figure 9A, the emulsion binder has high level of selectivity.

The highest ash rejection of 94.99% was achieved using 5 kg/t emulsion (Figure 9A). At 5 kg/t kerosene, 89.03% of the ash was rejected. The corresponding lowest ash content of 11.76%, achieved using 5 kg/t emulsion, indicates a high percentage rejection of inorganic minerals, including (but not limited to) silica (SiO<sub>2</sub>), kaolinite [Al<sub>2</sub>Si<sub>2</sub>O<sub>5</sub>(OH<sub>4</sub>)], calcite (CaCO<sub>3</sub>), and dolomite [CaMg(CO<sub>3</sub>)<sub>2</sub>]. This is because the emulsion has both hydrophobic and hydrophilic characteristics, leading to the possibility of hydrogen bonds forming between the carbon present in the coal sample and the HIP W/O emulsion binder (Lu *et al.*, 2019).

# Calorific value of the clean coal

The calorific value, also known as the heating value, was evaluated to determine the grade of the clean coal products obtained from the weathered coals, as well as their market potential. As a result of the rejection of the inorganic minerals of the clean coal produced at 5 kg/t W/O emulsion (Figures 7 and 9A), the feed coal calorific value increased from 20.94 MJ/kg to 28.61 MJ/kg (Figure 10). On the other hand, kerosene produced a clean product with a calorific value of 26.42 MJ/kg using the same dosage of 5 kg/t. As the optimal dosage of 5 kg/t (W/O emulsion binder) provided the best recovery from sample 1, the other coal samples (2, 3, 4, and 5) were then beneficiated at this condition.

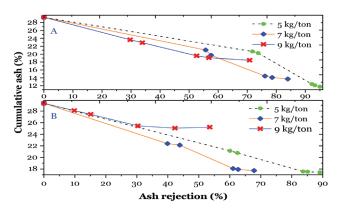


Figure 9—The rejection of inorganic minerals from reporting to the floated carbonaceous product at different HIP W/O (A) and kerosene (B) dosages

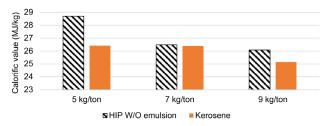


Figure 10—The calorific values of the clean coal products

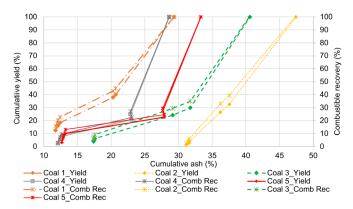


Figure 11—Effect of HIP W/O emulsion concentration (5~kg/ton) on the cumulative yield, combustible recovery (Comb Rec) and ash content of the clean coal products

# Recovery of coal sample 2, 3, 4, and 5 and with 5 kg/t of the emulsion

# Clean coal yield and combustible recovery

The recovery curves for the beneficiation of coals 2, 3, 4, and 5 at the same emulsion concentration (5 kg/t) to which coal 1 was subjected are depicted in Figure 11. The results were obtained by means of Equations [2] and [3]. Floated clean coal products with a combustible recovery of 2.5% and 6.14% and ash content of 31.16% and 17.23% were obtained for coals 2 and 3, respectively. The combustible recoveries for sample 4 and 5 were 3.08% and 4.02%, with ash contents of 12.12% and 12.70% respectively. For coal sample 1, a clean coal product with a recovery of 15.61% and 11.76% ash was achieved. The nature and degree of weathering of the coal could be responsible for the differences in clean coal product yield and ash content.

Table III					
Calorific values (MJ/kg) of the clean coal products using 5 kg/t emulsion binder					
Coal discards					
Value (MJ/kg)	1	2	3	4	5
Coal samples Clean coal CV increment	20.34 28.61 7.67	8.48 18.16 9.68	16.29 25.77 9.48	20.61 27.62 7.01	18.55 26.28 7.73
CV= calorific value	1	1			

# Impact of HIP W/O emulsion binder functional groups on selectivity

Table III presents the calorific values of the floated clean coal products with an emulsion concentration of 5 kg/t. As indicated, a higher calorific value increase was achieved for samples 2 and 3. The increase in the grade of both coals might be due to the high abundance of the OH groups found on their surfaces. According to Lu et al. (2019), HIP W/O emulsion is noted to prevent the entrainment of inorganic minerals in the floated clean coal product, and in addition, leads to the formation of strong bonds with the oxygenated functional groups in the weathered coal. The highest calorific value improvement was achieved for sample 2, from 8.48 MJ/kg to 18.26 MJ/kg after beneficiation. This calorific value surpasses the heating value requirements for some Eskom power plants such as Lethabo power station, which requires a calorific value of 16 MJ/kg and ash content of between 35% and 42% ash (Eskom, n.d.). On the other hand, the clean coal obtained from coals 1 and coal 4 had a calorific value of 28.61 MJ/kg, and 27.62 MJ/kg and can be classified as Grade A and Grade B, respectively. As shown in Table III, the products obtained from samples 3 and 5 have heating values of 25.77 MJ/kg and 26.28 MJ/kg, respectively.

# Conclusions

In summary, the HIP W/O emulsion binder has the capacity to beneficiate weathered coal samples from tailings dams. Compared to pure kerosene, requiring 5 kg/t to produce clean coal products, the HIP W/O emulsion binder used only 0.25 kg/t of the organic liquids (kerosene and Span 80) to produce a clean coal concentrate of 11.76% ash content and heating value of 28.61 MJ/kg. In contrast, 17.47% ash and a calorific value of 26.42 MJ/kg were achieved using 5 kg/t of kerosene. This research has established that weathered coal samples from South African coalfields dumped for decades can be upgraded to a high-quality product.

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